

**zumdaHL chemistry 8th edition pdf**

An acid is a molecule or ion capable of donating a hydron (proton or hydrogen ion H<sup>+</sup>), or, alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids is the proton donors or Brønsted acids. In the special case of aqueous solutions, proton donors form the hydronium ion H<sub>3</sub>O<sup>+</sup> and are known as Arrhenius acids.

**Acid - Wikipedia**

Propiedades de los Ácidos. Tienen sabor agrio como en el caso del Ácido cítrico en la naranja y el limón.; Cambian el color del papel tornasol azul a rosa, el anaranjado de metilo a rojo y deja incolora a la fenolftaleína.; Son corrosivos.

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